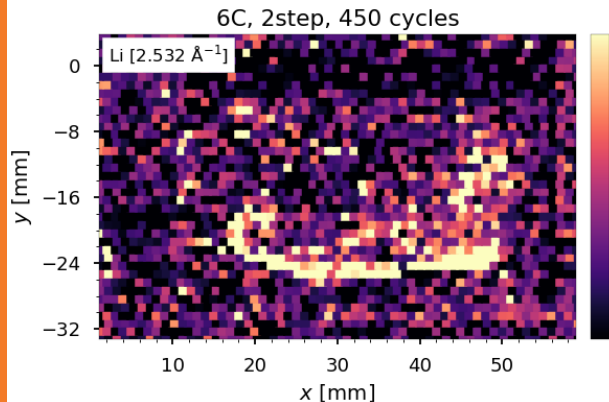


BAT338



eXtreme Fast Charge Cell Evaluation
of Lithium-ion Batteries

EXTREME FAST CHARGING CELL DEVELOPMENT OVERVIEW



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Argonne National Lab

SAMUEL GILLARD
Department of Energy

This presentation does not contain any proprietary,
confidential, or otherwise restricted information

OVERVIEW

Timeline

- Start: October 1, 2017
- End: September 30, 2021
- Percent Complete: 37%

Barriers

- Cell degradation during fast charge
- Low energy density and high cost of fast charge cells

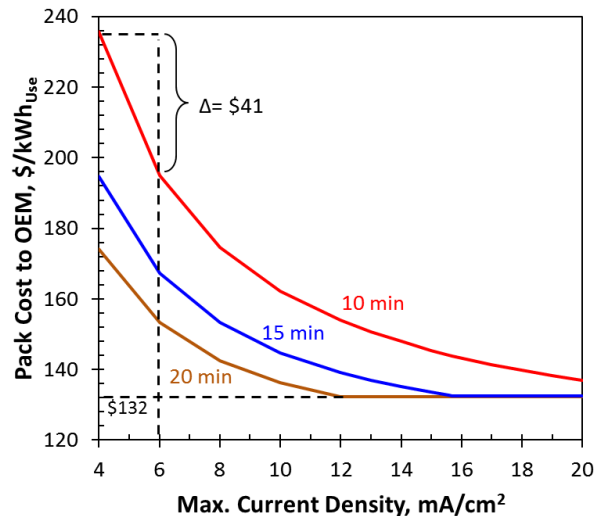
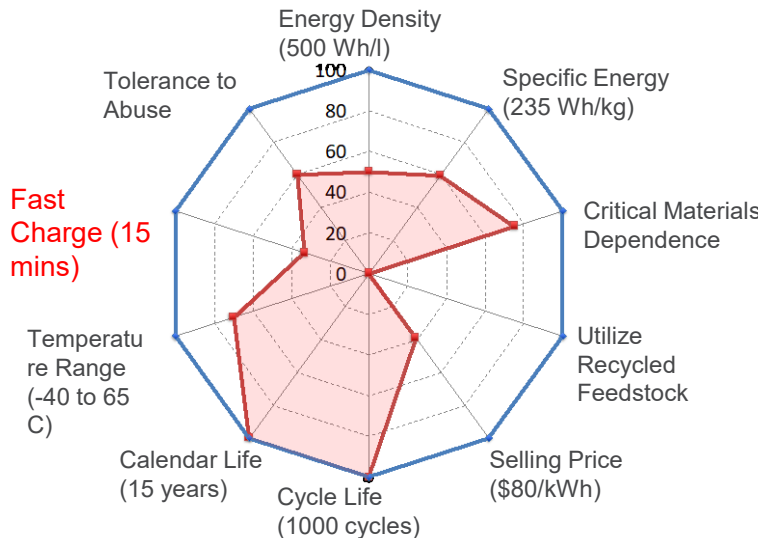
Budget

- Funding for FY19 – 6390k
 - ANL – 2400k
 - NREL – 1600K
 - INL – 440K
 - SLAC – 1000K
 - LBNL – 950K

Partners

- Argonne National Laboratory
- Idaho National Laboratory
- Lawrence Berkeley National Lab
- National Renewable Energy Laboratory
- ²SLAC National Accelerator Lab

RELEVANCE: FAST CHARGE REMAINS AN ISSUE FOR WIDESPREAD ADOPTION OF EVS



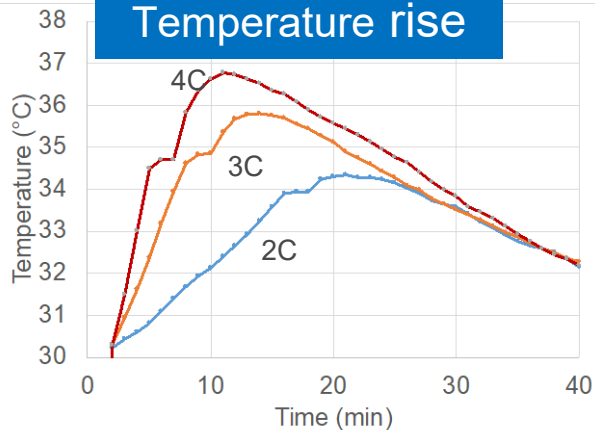
Today

Future

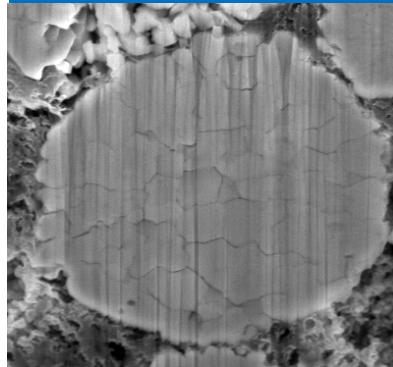
Fast charging a major issue. While fast charge cells exist, they are cost prohibitive or have poor life

RELEVANCE: WHAT LIMITS FAST CHARGE?

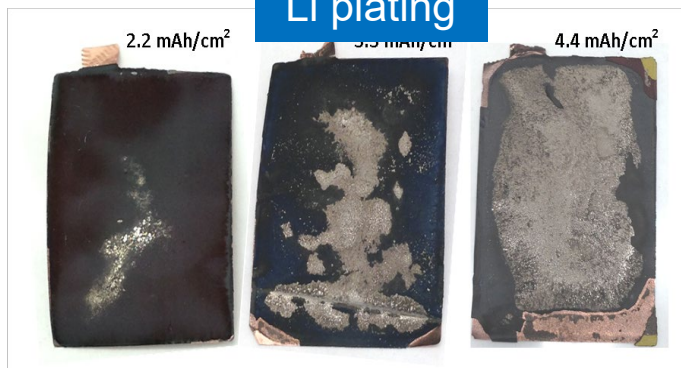
Temperature rise



Particle breakup

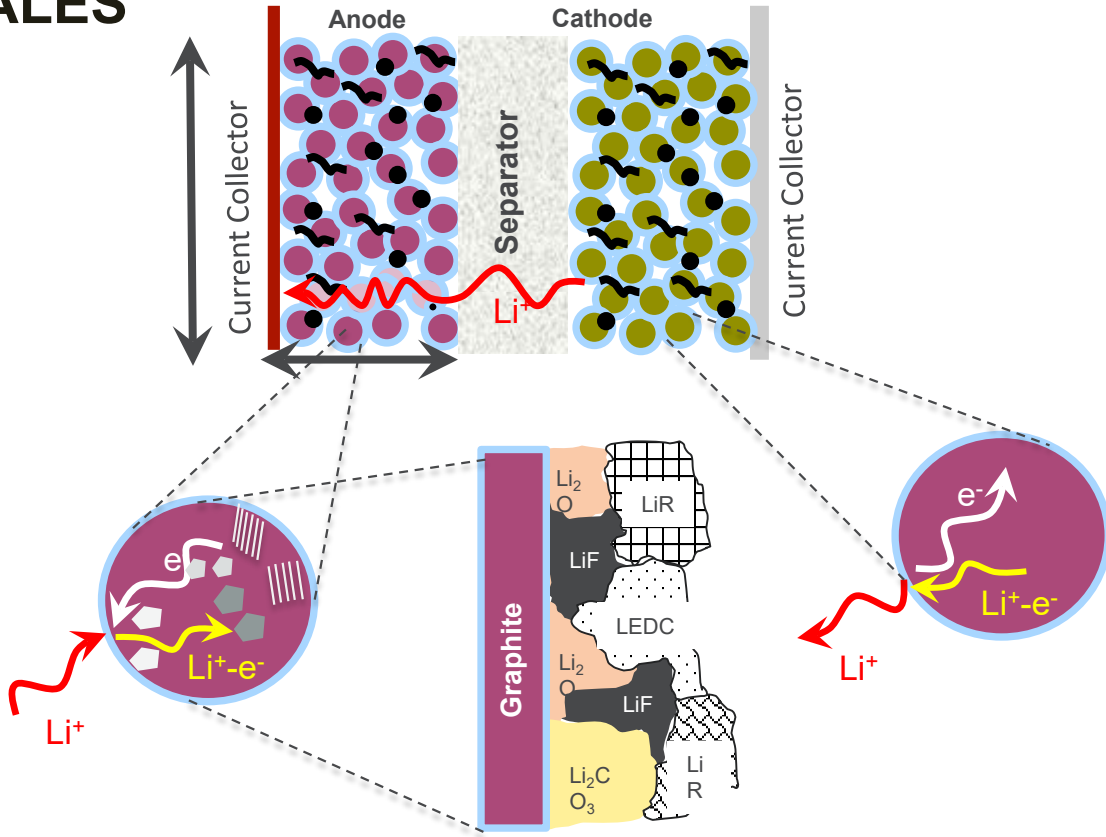


Li plating



Li plating remains a critical issue

RELEVANCE: CHALLENGES AT MULTIPLE SCALES



QUESTIONS TO HELP MAP THE PROBLEM

1. How does the anode design (loading, porosity...) influence the propensity for Li plating? **BAT339**
2. How does the charging protocol and extended cycling impact degradation (anode vs. cathode) during fast charge? **BAT340**
3. Can we quantify the bottleneck at the electrode scale? Can we start to examine solutions to enable fast charge? **BAT371**
4. How important are local heterogeneities in determine the propensity for Li plating? Does Li transport in graphite depend on charging rates? **BAT383**
5. Can we detect Li plating *in situ*? And at what scale? **BAT384**

COLLABORATION ACROSS LABS AND UNIVERSITIES



Cell and electrode design and building, performance characterization, post-test, cell and atomistic modeling, cost modeling



Li detection, electrode architecture, diagnostics



Performance characterization, failure analysis, electrolyte modeling and characterization, Li detection



Thermal characterization, life modeling, micro and macro scale modeling, electrolyte modeling and characterization



Li detection, novel separators, diagnostics



CONTRIBUTORS AND ACKNOWLEDGEMENTS

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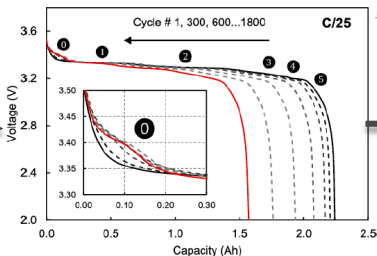


*Support for this work from the Vehicle Technologies Office,
DOE-EERE – Samuel Gillard, Steven Boyd, David Howell*



APPROACH: COMBINE CUSTOM DESIGNED ELECTRODES WITH ADVANCED CHARACTERIZATION AND MULTISCALE MODELING

Leverage the strengths among the partner institutions



Cells with different designs

- Different graphite's
- Different loading
- Cells with and without reference electrode

Fast charge testing and electrochemical diagnostics

- Different charge rates.
- Short vs. long term cycling
- Coulombic efficiency
- Electrochemical signatures of plating

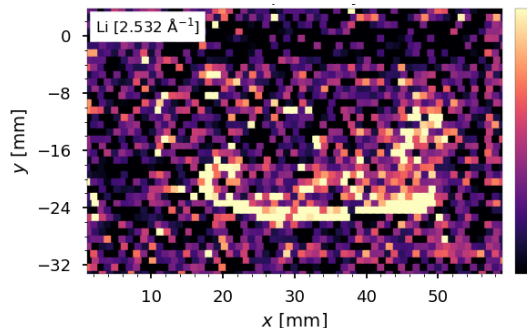
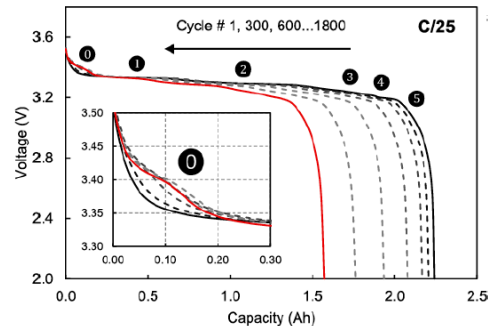
Characterization: in situ and post test.

- X-ray-based: reaction distribution
- Li detection: SEM, XPS.
- SEI detection: Raman, FTIR, XPS, HPLC
- Particle cracking: XRD, SEM

DETECTING LI PLATING: A CHALLENGE

In situ methods. Question: How close to nucleation can we detect plating?

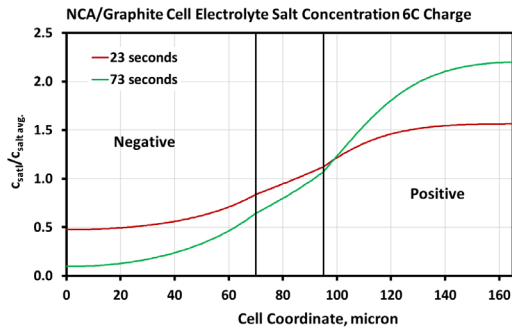
- Electrochemical (coulombic efficiency, capacity fade, electrochemical signatures)
- Reference electrode methods
- Acoustic
- Thermal
- X-ray-based analysis
- Gas evolution
- ...



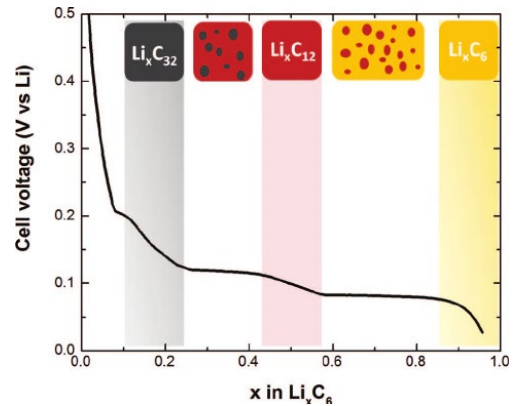
Detecting Li plating in situ at its initiation remains a key challenge

MODELING AT DIFFERENT SCALES TO IDENTIFY LIMITATIONS

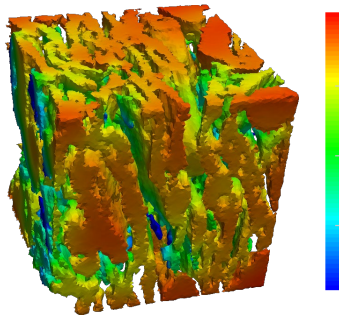
Electrode level effects



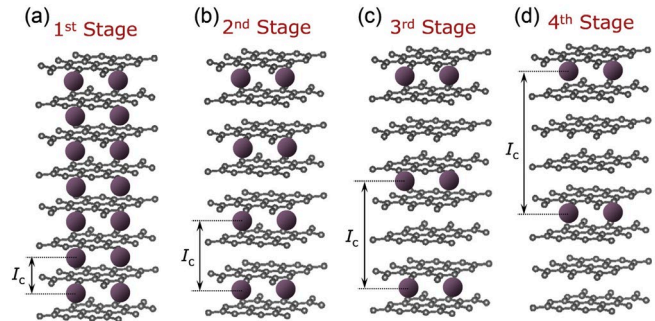
Particle-scale issues



Microstructure scale effects

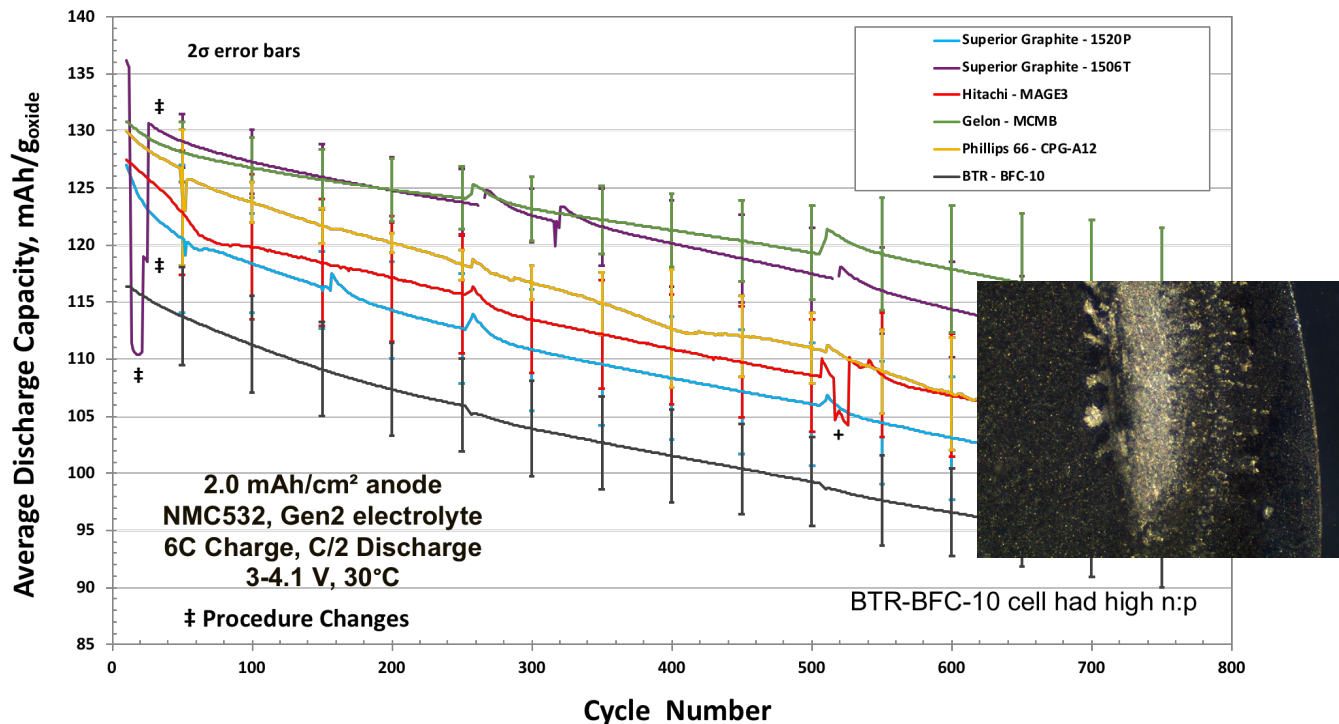


Graphite layer scale



ANODES CYCLE WELL...EVEN WHEN THEY PLATE

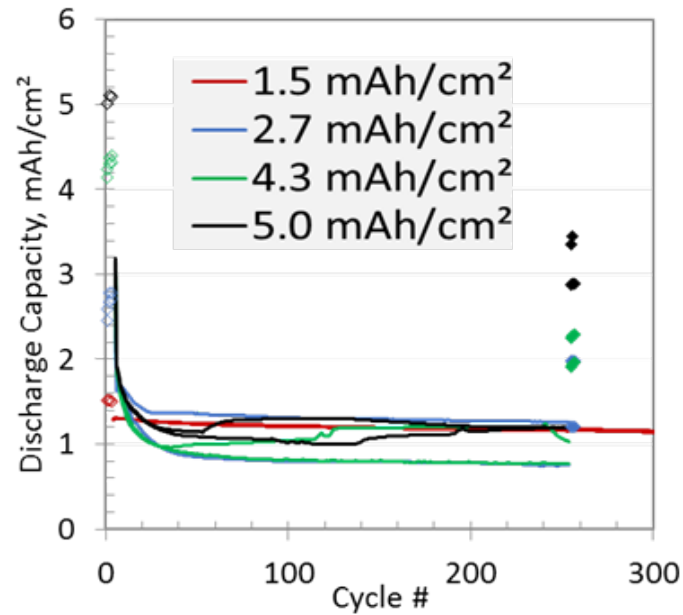
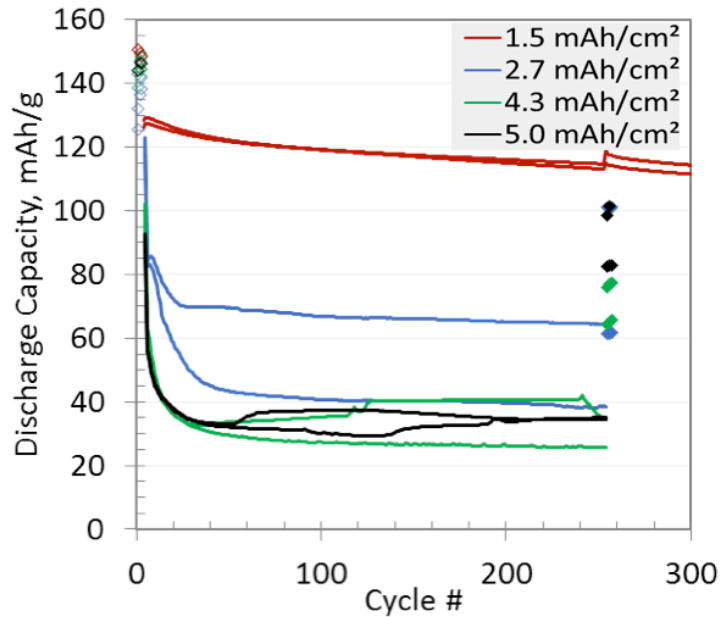
6 C charge on 2 mAh/cm² anode cells



Thin cells cycle well. And fade seems similar with different graphite's and on CMC-SBR and PVDF binder cells

WHAT HAPPENS WHEN WE INCREASE THE LOADING?

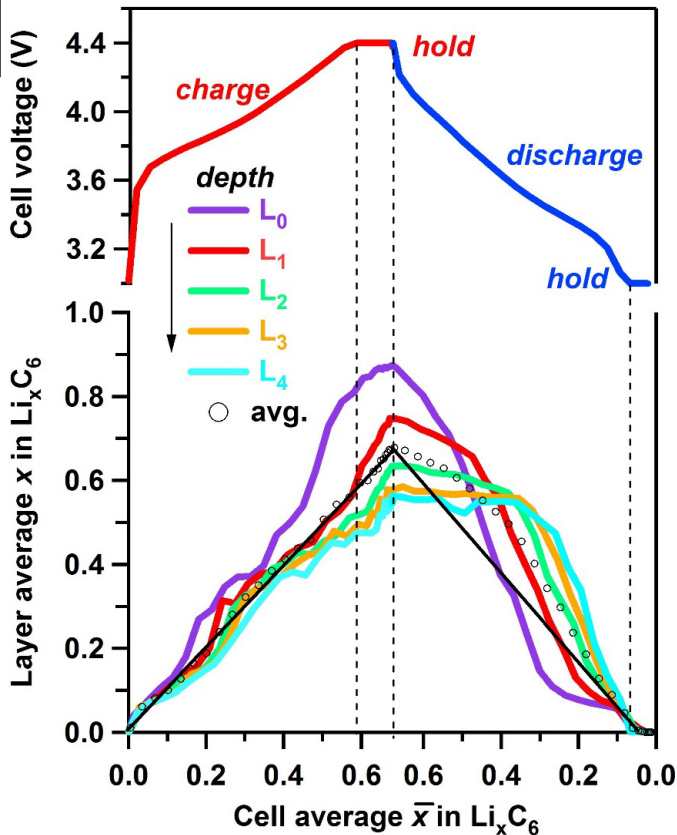
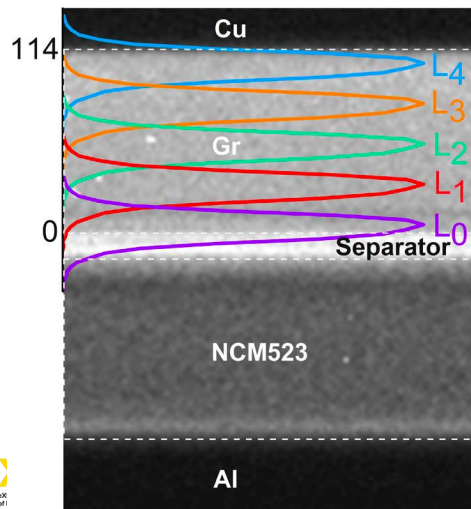
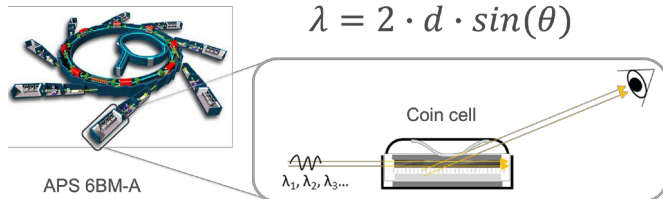
Conoco Philips GCP-A12



Results suggest the need to examine the reaction distribution in the porous anode

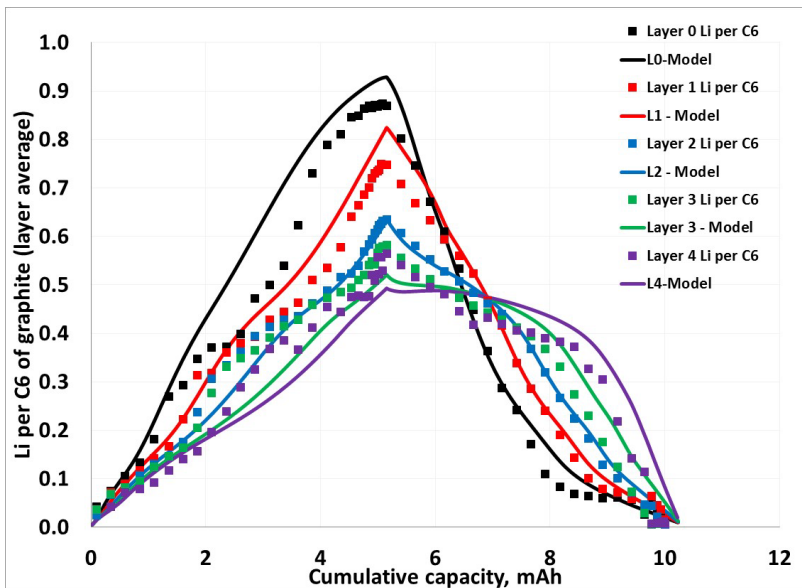
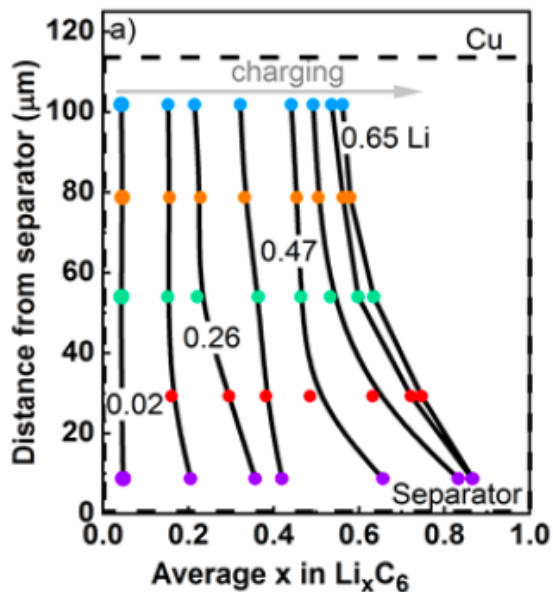
IN SITU SYNCHROTRON EDXRD DETECTION OF REACTION DISTRIBUTION

Dan Abraham, *Energy & Environmental Science* 12.2 (2019): 656-665.



MODELS SHOW REASONABLE PREDICTIVE CAPABILITY

1 C rate

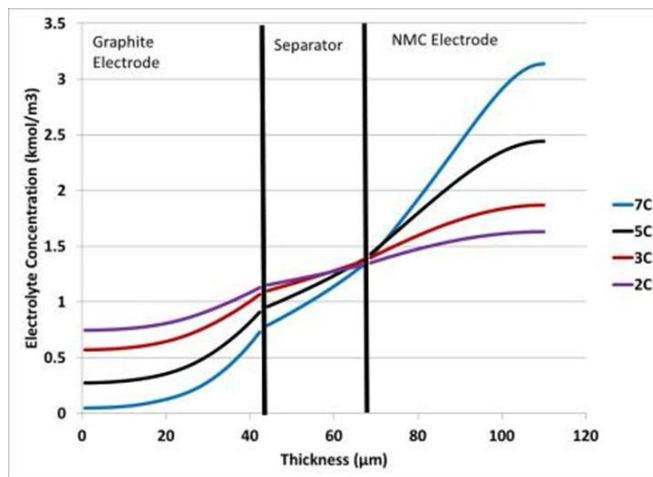


MODEL PROVIDES INSIGHTS INTO OBSERVATIONS

Low Electrode Loading

1.5 mAh/cm² cathode (42 μm)

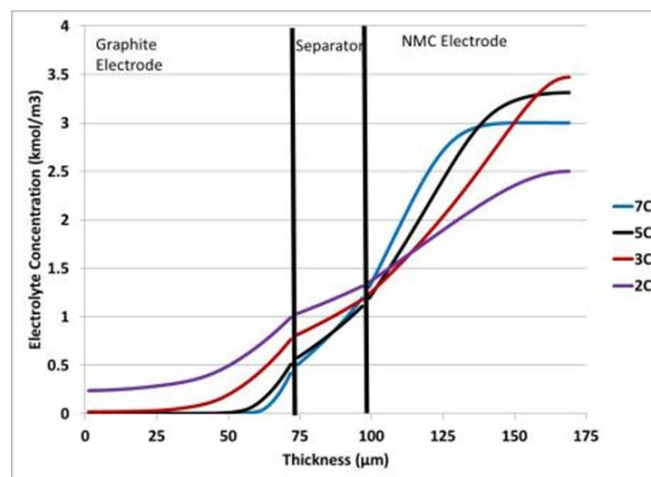
1.84 mAh/cm² anode (43 μm)



Medium Electrode Loading

2.5 mAh/cm² cathode (71 μm)

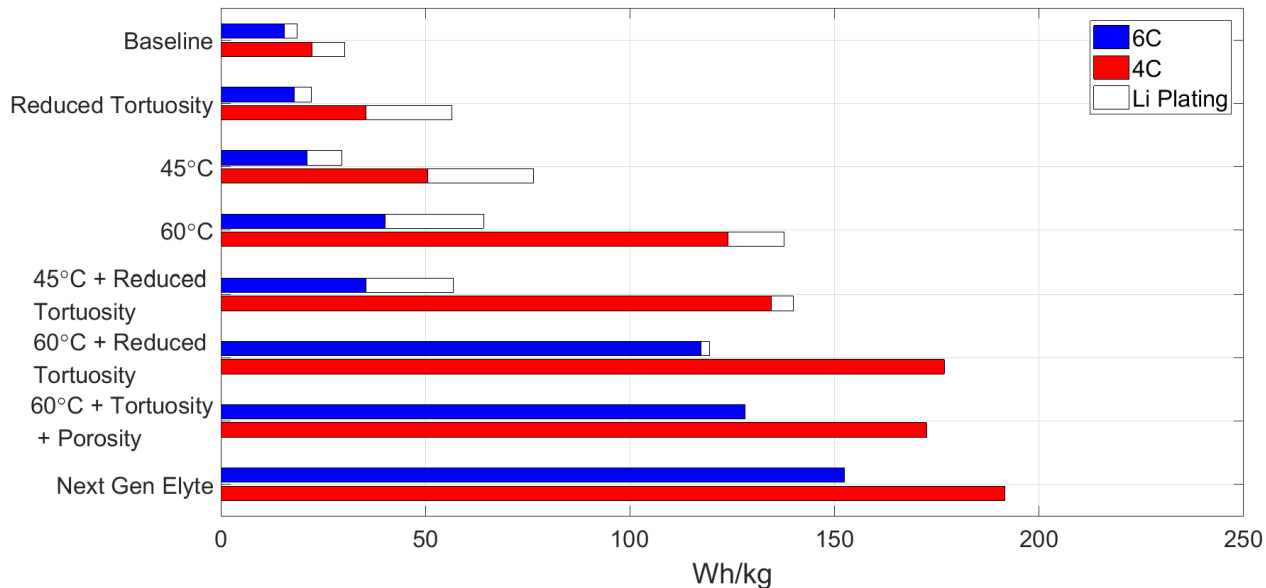
3.07 mAh/cm² anode (87 μm)



Fast charging will require enhancing the electrolyte transport in the porous electrode

MODELS SHOW IMPACT OF DIFFERENT APPROACHES

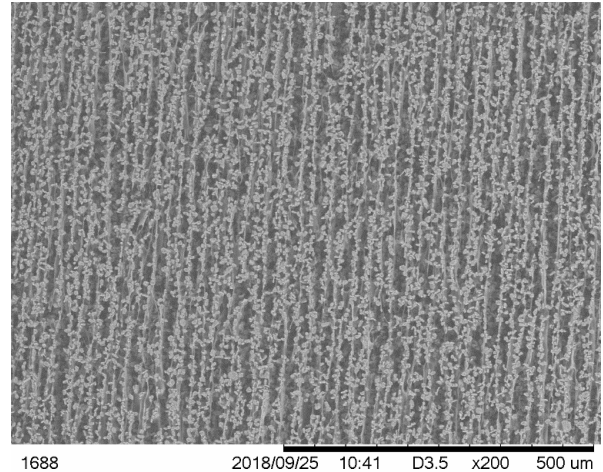
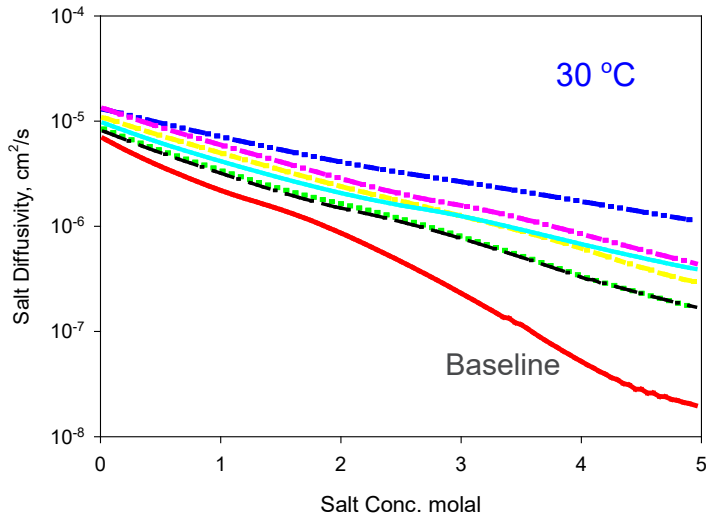
4 mAh/cm² (230 Wh/kg cell; 110 micron electrodes)



- Next Gen Electrolyte = 2X ionic conductivity, 4X diffusivity, and transference number increased by 0.15

Need a combination of approaches

WE ARE NOW EMBARKING ON TESTING THESE PREDICTIONS

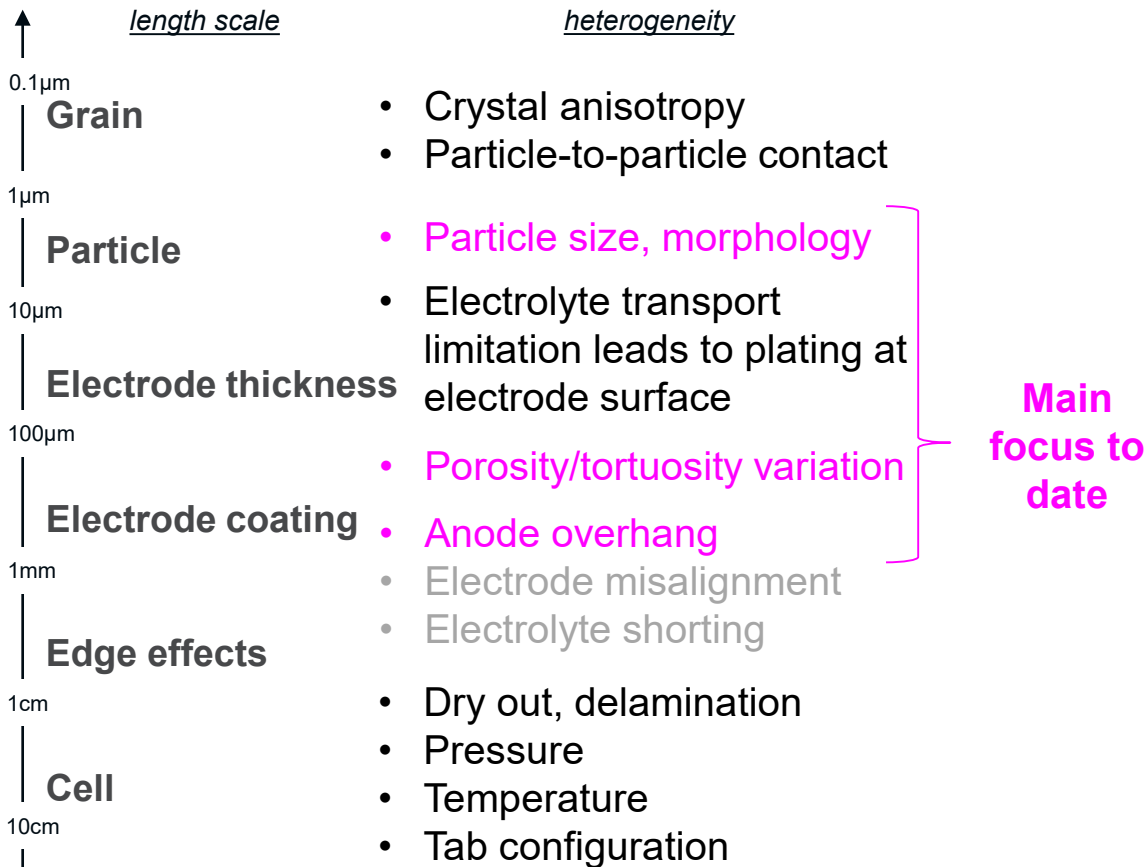


Higher transport property
electrolytes

Reducing tortuosity in the
anode

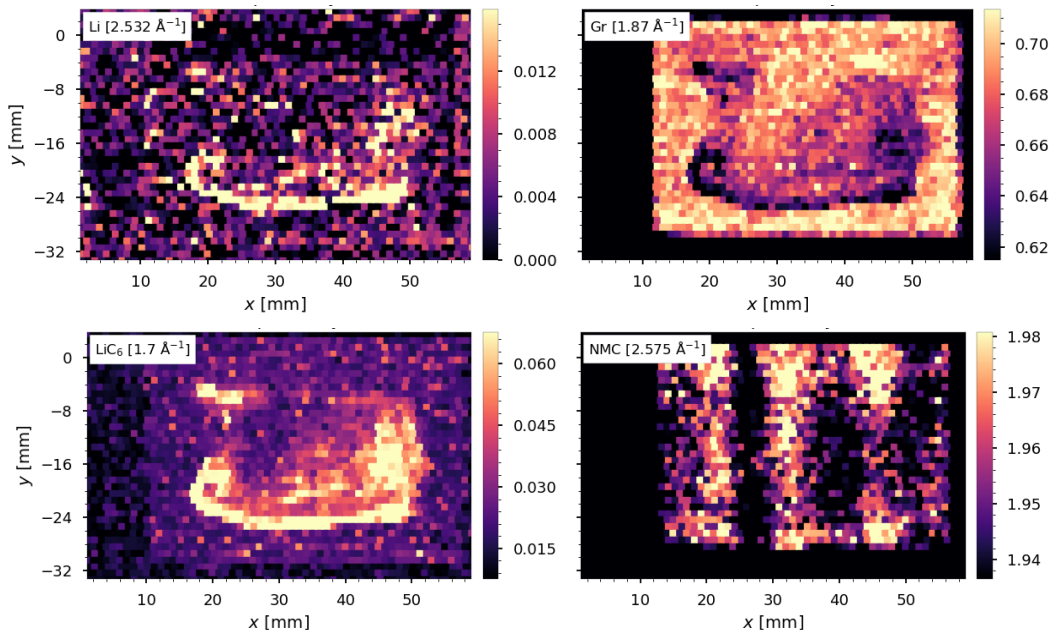
HETEROGENEITIES BECOMING A FOCUS AREA

Heterogeneities at all length scales cause early onset of Li plating



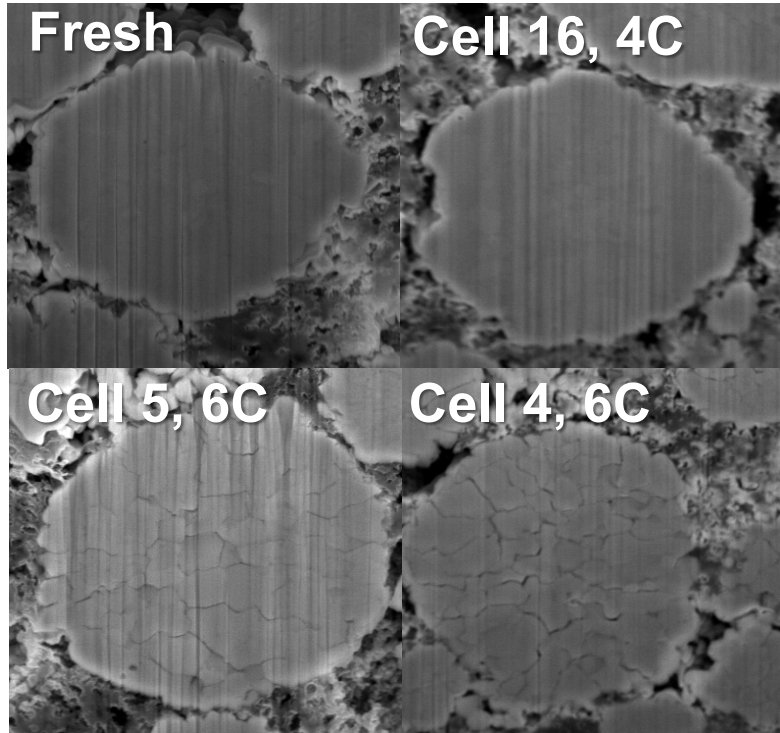
Li DETECTION EFFORTS PROVIDING INSIGHTS

Cycling: 6C, CCCV, 450 cycles



- Intensities of Li and Gr anti-correlated
- Intensities of Li and LiC₆/LiC₁₂ correlated
- NMC shows pattern; no obvious correlation with Li

WE ARE ALSO SEEING EARLY EVIDENCE OF CATHODE CRACKING



REMAINING CHALLENGES AND BARRIERS

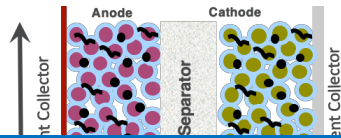
Will be discussed as part of BAT386

PROPOSED FUTURE RESEARCH

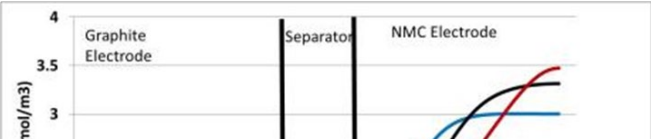
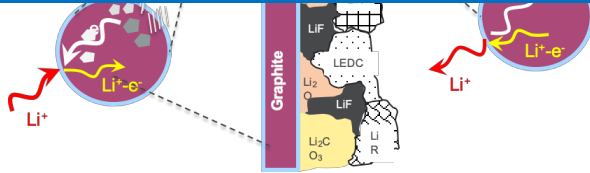
Will be discussed as part of BAT386

Any proposed future work is subject to change based on funding levels.

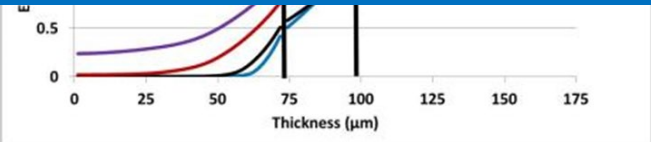
SUMMARY



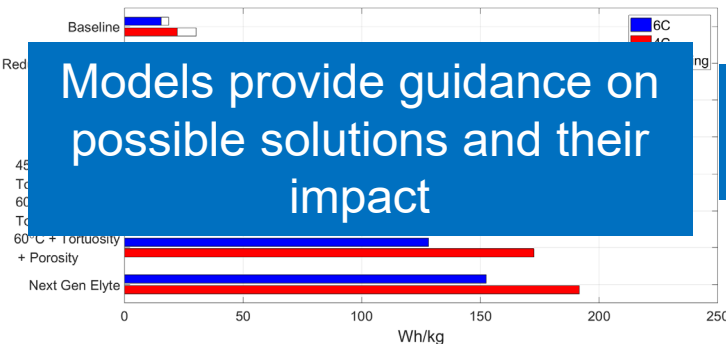
Fast charging limited by processes at many scales



Reaction zone concentrated at the separator interface leads to Li plating



Models provide guidance on possible solutions and their impact



Cell 5, 6C

Cathode fade starting to come to focus

